

Statistical Mechanics — PHY 5524

Midterm March 2, 2009

1. (60 pts) Consider a system of  $N$  noninteracting spin-1 nuclei placed in a magnetic field  $B$  at temperature  $T$ . The nuclei form a solid so you may assume that the spins are localized (i.e. distinguishable). The Hamiltonian for each spin is

$$H = -\mu Bs,$$

where  $\mu$  is the magnetic moment strength and  $s = S_Z/\hbar$  can take the values  $-1, 0,$  and  $1$ .

(a) Write down the partition function for this system and obtain the Helmholtz free energy.

(b) Determine the magnetization of this system  $M = N\mu\langle s \rangle$  as a function of  $N, B$  and  $T$ .

(c) Consider the limit  $k_B T \gg \mu B$  and obtain an approximate expression for  $M$ , valid to first order in  $\frac{\mu B}{k_B T}$ . Using this expression determine the magnetic susceptibility of the nuclei,

$$\chi(T) = \lim_{B \rightarrow 0} \left( \frac{\partial M}{\partial B} \right)_{N,T}.$$

(d) Now determine the entropy of this system. From your result show that for adiabatic processes  $B$  and  $T$  obey a law of the form  $T^\alpha B = \text{Const.}$  and determine the exponent  $\alpha$ .

2. (40 pts) Consider a classical gas consisting of  $N$  noninteracting particles confined to a volume  $V$  at temperature  $T$ . Assume the partition function for a single particle in this gas has the form

$$Q_1 = Vf(T),$$

for some (unspecified) function  $f(T)$ .

(a) Write down the partition function for this gas of  $N$  particles and obtain an expression for the Helmholtz free energy  $A$ . Comment on the significance of the Gibbs factor in determining whether or not your expression for  $A$  is extensive.

(b) From the Helmholtz free energy determine the pressure  $P$  of this gas and show that it satisfies the ideal gas law regardless of the form of  $f(T)$ .

(c) Now assume that  $f(T) = \gamma T^n$  for some constant  $\gamma$  and  $n$ . Obtain the energy of the gas  $E$ , and from this determine  $C_V$ , the specific heat at constant volume, and  $C_P$ , the specific heat at constant pressure.